

Condensed Matter Physics I

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Exercises - Drude Model

Hint: normally the Ashcroft-Mermin book uses CGS units; however, it can be convenient to use MKS(SI) units in presence of eqs. containing the resistivity (use ρ in Ω -meters, m in Kg, n in electrons/ m^3 , e in Coulomb).

Exercise 1

1. Consider typical values of the magnetic field ($H(\text{Earth}) \approx 0.5$ Gauss; $H(\text{lab}) \approx 1-10$ T). Evaluate the product $\omega_c \tau$, where ω_c is the cyclotron frequency and τ the electron relaxation time. On average, can an electron in a metal make many revolutions between collisions or not?

Exercise 2

1. Give a numerical estimate of the mean electronic velocity in case of a current density of 0.1 A/ mm^2 flowing in a copper wire ($n_{el} = 8.47 \cdot 10^{22}$ cm^{-3}). [$v = 0.007$ cm/s]

Exercise 3

Consider Al at room temperature. Its electron density is $n = 18.1 \cdot 10^{22}/\text{cm}^3$ and its electrical resistivity is $\rho = 2.45 \mu\Omega \cdot \text{cm}$.

1. Find its electron relaxation time τ and electron mean free path ℓ in the Drude model. [$\tau = 8.01 \cdot 10^{-15}$ s; $\ell = 13 \text{ \AA}$]
2. Consider AC conductivity. At which frequency ω the real part of the conductivity $\sigma(\omega)$ will be 1/10 of its zero-frequency value? [$\nu = \omega/2\pi = 59.7 \cdot 10^{12}$ THz]

Exercise 4

Sodium (Na) in standard temperature and pressure conditions is a metal with BCC structure, density of about 0.97 g cm^{-3} and mass number = 23.

1. Calculate the atomic density (number of atoms per unit volume) of solid sodium. [$n_{at} = 2.54 \cdot 10^{22}$ cm^{-3}]
2. Calculate the electron density. [$n_{el} = n_{at}$]
3. Calculate the plasma frequency. [$\omega_p = 8.99 \cdot 10^{15}$ s^{-1}]
4. If the electron gas is treated as a classical one, which is the kinetic energy at $T = 0^\circ$ K? and at room temperature?
5. Given the resistivity in DC at room temperature, $\rho = 4.2 \mu\Omega \text{ cm}$, calculate the relaxation time τ . [$\tau = 3.3 \cdot 10^{-14}$ s]

Other exercises

Ex. 2 and 3 of Chapter 1 of A&M